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Photoelectrochemical Characterisation of ZnSe coated Copper Indium Sulphide Thin Film Electrodes

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Copper Indium Sulphide thin films were prepared by sulphidation of Cu-In alloy on Ti substrate. Cu-In alloy was potentiostatically electrodeposited at -1.4 V Vs SCE in an aqueous bath containing 5 mM CuCl₂, 37.5 mM InCl₃, 1% (V/V) TEM and .75% (V/V) ammonia.

Sulphidation was carried out in saturated H_2S gas at 550 °C for 30 min. XRD measurement reveals that the crystal structure of the films is $CuIn_{11}S_{17}$. ZnSe was deposited on CuInS electrode by electrodeposition in an aqueous bath of 0.1 M ZnSO₄ and 10^{-5} M SeO₂ at -0.5 V Vs SCE for 90 min. XRD measurement reveals that the ZnSe films are amorphous.

Ti/CuInS/ ZnSe thin film system in a PEC cell containing KI produces n-type photocondutivity. Dark and illuminated I-V measurement shows the existence of a main junction. However, some departure is also evident suggesting the possibility of existence of another junction. Spectral response of the Ti/CuInS/ZnSe system in a PEC cell shows the photoresponse of both n-CuInS/p-ZnSe and p-ZnSe/electrolyte interfaces. Normally, for shorter wavelength the spectral response is p-type while for the long wavelength it is n-type.

The preliminary results of the study suggest the possibility of utilising electrodeposited n-CuInS in combination with electrodeposited p-ZnSe in developing a low-cost thin film solar cell.

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