

## **Fabrication of an efficient $\text{Cu}_2\text{O}$ homojunction by electrodeposition technique for solar cell applications**

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### **Abstract**

Electrodeposition is a very attractive low cost technique for fabrication of  $\text{Cu}_2\text{O}$  homojunction solar cells. Although electrodeposited p-n homojunction  $\text{Cu}_2\text{O}$  (metal substrate/p- $\text{Cu}_2\text{O}$ /n- $\text{Cu}_2\text{O}$ ) solar cells were reported earlier, n-p homojunction  $\text{Cu}_2\text{O}$  (metal substrate/n- $\text{Cu}_2\text{O}$ /p- $\text{Cu}_2\text{O}$ ) solar cells are very limited in the literature. This solar cell structure is very important when exploring the possibilities to improve the efficiencies of reported  $\text{Cu}_2\text{O}$  homojunction solar cells. In this study, current-voltage characteristics and spectral response measurements were employed to investigate the possibilities of fabrication of n-p homojunction  $\text{Cu}_2\text{O}$  solar cell by electrodeposition technique. Different deposition conditions were adopted to grow and optimize the p-type and n-type  $\text{Cu}_2\text{O}$  films. n- $\text{Cu}_2\text{O}$  thin films were electrodeposited on Ti substrate using an acetate bath of pH 6.1, where the resulted films produced only the n-type photoresponse in a PEC. Subsequently, p- $\text{Cu}_2\text{O}$  thin film was electrodeposited on Ti/n- $\text{Cu}_2\text{O}$  electrode using an acetate bath with a cupric ion concentration of 0.001 M. This study revealed the possibility of fabrication of an efficient n-p homojunction of  $\text{Cu}_2\text{O}$  for the applications in solar cells by consecutive electrodeposition of an n- $\text{Cu}_2\text{O}$  film followed by a p- $\text{Cu}_2\text{O}$  film using an acetate bath.

### **1.0 Introduction**

Cuprous oxide ( $\text{Cu}_2\text{O}$ ) is an attractive material for photovoltaic applications due to its unique properties [1-6]. It is a defect type semiconductor and it is well established as a p-type material due to the Cu vacancies created in the crystal lattice [7-12]. However, it has been reported earlier that n- $\text{Cu}_2\text{O}$  films can be deposited using the electrodeposition technique [13]. Origin of the n-type conductivity of  $\text{Cu}_2\text{O}$  is considered as due to the excess of Cu ions and/or O vacancies created in the  $\text{Cu}_2\text{O}$  lattice. In general, conductivity type in electrodeposited  $\text{Cu}_2\text{O}$  films strongly depends on the pH and cupric ion concentration of the depositing bath solution [14-15]. Acidic baths produce n-type films while basic baths produce p-type films. However, electrodeposition technique of  $\text{Cu}_2\text{O}$  is very attractive because of its simplicity, low cost and low-temperature process. Indeed, control of deposition parameters of the bath to produce n- $\text{Cu}_2\text{O}$  or p- $\text{Cu}_2\text{O}$  thin films having better optoelectrical properties is very important for them to be used in solar cell applications.

Many authors have reported the possibility of the p-n homojunction  $\text{Cu}_2\text{O}$  (metal substrate/p- $\text{Cu}_2\text{O}$ /n- $\text{Cu}_2\text{O}$ ) solar cells [14, 16-18]. However, to our knowledge, fabrication of n-p homojunction  $\text{Cu}_2\text{O}$  (metal substrate/n- $\text{Cu}_2\text{O}$ /p- $\text{Cu}_2\text{O}$ ) solar cells using only acetate bath has not been reported earlier although Jayathilaka et al [19] has reported the possibility of fabrication of similar solar cell using acetate and lactate baths for growth of n- $\text{Cu}_2\text{O}$  and p- $\text{Cu}_2\text{O}$  respectively. The reason may be that during the growth of p-type